

Name: \_\_\_\_\_

## Balancing Equations Notetaker

Law of Conservation of Mass: \_\_\_\_\_

\_\_\_\_\_  
\_\_\_\_\_

Because of the Law of Conservation of Mass an equation must be balanced and have the same number of atoms \_\_\_\_\_

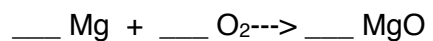
\_\_\_\_\_

Reactants: \_\_\_\_\_

Products: \_\_\_\_\_

Reactant + Reactant = Product

Let's try one!



### Balancing Equations an Easier Way!

Step 1: Write the formula

Step 2: List the atoms involved on each side

Step 3: Start balancing

Step 4: Adjust totals until everything is balanced

Name: \_\_\_\_\_

Try Balancing the equations below!

